

## ERRATA

L. P. FILIPPOV: Research of liquid thermal conductivity at Moscow University, *Int. J. Heat Mass Transfer* **11**, 331-345 (1968).

The printer regrets that in Table 2 (p. 337) some figures have been misplaced. The table should begin as follows:

Table 2

Formula	Substances	Temperature range		kcal/mh degC	$\alpha \times 10^3$ deg C <sup>-1</sup>
		lower	upper		
CCl <sub>4</sub>	carbon tetrachloride	15	90	0.087 <sub>5</sub>	2
CHBr <sub>3</sub>	bromoform	10	90	0.087	2.1
CHCl <sub>3</sub>	chloroform	15	70	0.101	1.8
CH <sub>2</sub> O <sub>2</sub>	formic acid	15	90	0.23 <sub>5</sub>	—
CH <sub>4</sub> O	methanol	15	90	0.175	1
C <sub>2</sub> H <sub>4</sub> Br <sub>2</sub>	ethylen bromide (1.2-ethylene dibromide)	15	90	0.086 <sub>5</sub>	1.3
C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	ethylen dichloride (1.2-ethylene dichloride)	15	80	0.115 <sub>5</sub>	1.5
C <sub>2</sub> H <sub>4</sub> O	acetaldehyde	10	30	0.159 <sub>5</sub>	2.6
C <sub>2</sub> H <sub>3</sub> ClO <sub>2</sub>	monochloroacetic acid (supercooled)	40	90	0.145	non-linear
C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	acetic acid	5	90	0.143	1
C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	methyl-formate	10	40	0.163	1.2
C <sub>2</sub> H <sub>5</sub> Br	ethyl-bromide	15	45	0.087 <sub>5</sub>	2
C <sub>2</sub> H <sub>5</sub> I	ethyl-iodide	20	70	0.077 <sub>5</sub>	1.0
C <sub>2</sub> H <sub>5</sub> ON	acetamide	65	100	0.216	~0
C <sub>2</sub> H <sub>5</sub> O	ethanol	15	90	0.140	1.9
C <sub>2</sub> H <sub>6</sub> O <sub>2</sub>	ethylen-glycol	15	90	0.220	-1.0
C <sub>3</sub> H <sub>6</sub> O	acetone	15	50	0.139 <sub>5</sub>	2.2
C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	ethyl-formate	15	50	0.139	(3)
C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	methyl-acetate	10	50	0.134	2
C <sub>3</sub> H <sub>7</sub> O	propanol	15	80	0.13	1.4
C <sub>3</sub> H <sub>7</sub> O	i-propanol	10	80	0.120	1
C <sub>3</sub> H <sub>7</sub> O <sub>3</sub>	glycerine	15	90	0.244	-2.3
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	acetaldehyde	15	90	0.140 <sub>5</sub>	1.5
C <sub>4</sub> H <sub>7</sub> O <sub>2</sub>	butyric acid	10	90	0.128	0.9
C <sub>4</sub> H <sub>7</sub> O <sub>2</sub>	ethyl-acetate	15	90	0.123 <sub>5</sub>	2
C <sub>4</sub> H <sub>7</sub> O <sub>2</sub>	propyl-formate	15	70	0.124 <sub>5</sub>	1.7
C <sub>4</sub> H <sub>10</sub> O	butanol	15	90	0.123	1.4
C <sub>4</sub> H <sub>10</sub> O	i-butanol	15	90	0.115	1.0
C <sub>4</sub> H <sub>10</sub> O	diethylether	10	70	0.114	2.3

J. W. ROSE: Condensation of a vapour in the presence of a non-condensing gas, *Int. J. Heat Mass Transfer* **12**, 233-237 (1969).

In equation (8) on p. 234 ( $W^0$  should read  $W_0$ ).